

Supplement of Atmos. Meas. Tech., 10, 1445–1463, 2017
<http://www.atmos-meas-tech.net/10/1445/2017/>
doi:10.5194/amt-10-1445-2017-supplement
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Supplement of

Measurement of alkyl and multifunctional organic nitrates by proton-transfer-reaction mass spectrometry

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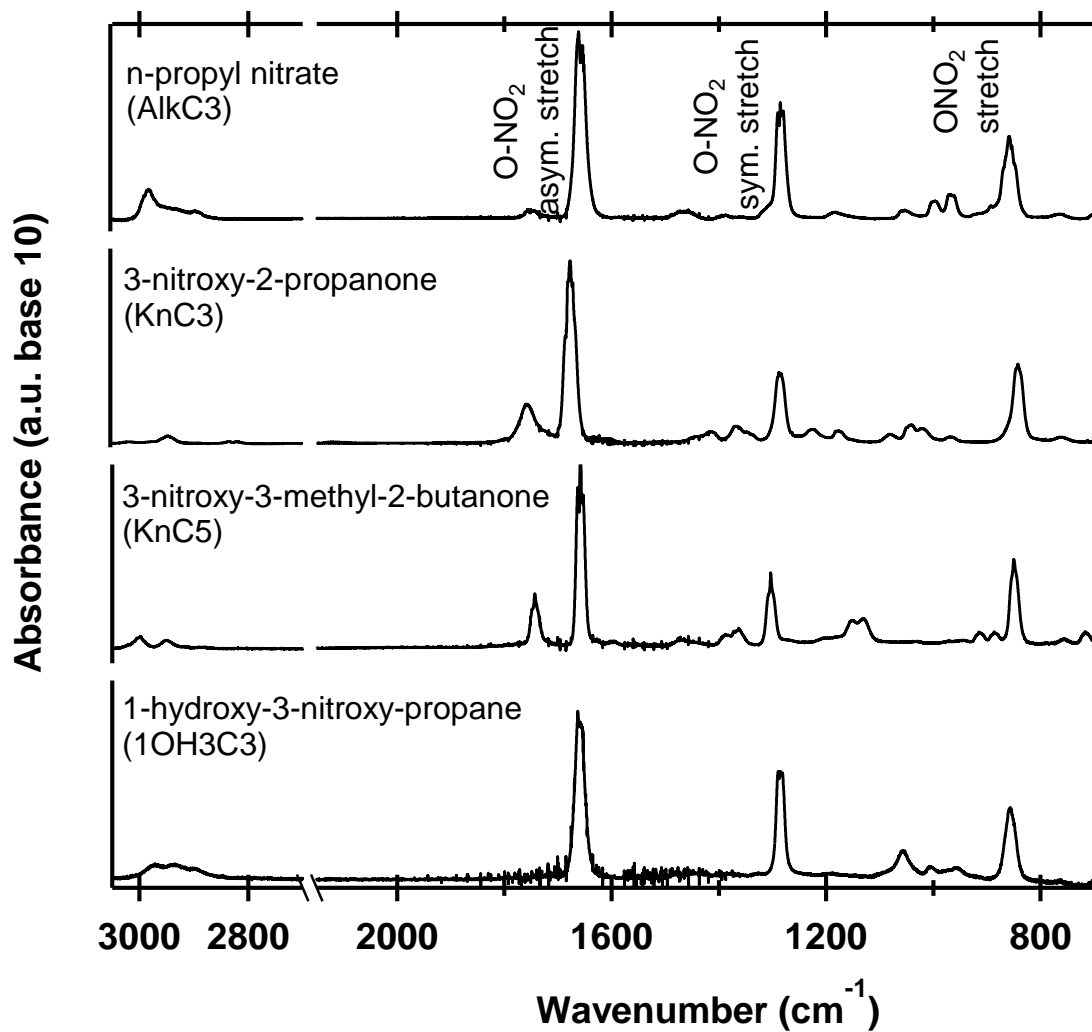


Figure S1. The FTIR gas phase absorption spectra of the synthesized organic nitrates (KnC3, KnC5, 1OH3C3) compared with a commercially available alkyl nitrate spectra (AlkC3).

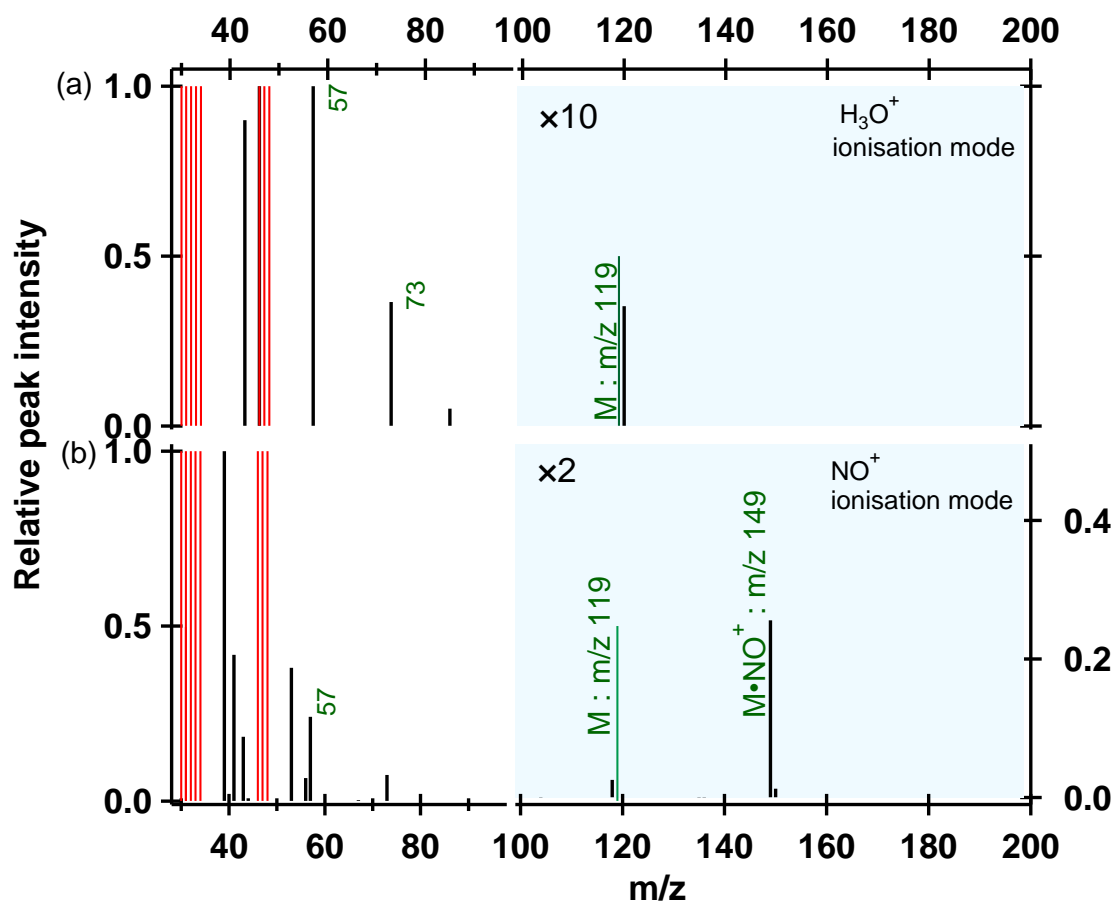


Figure S2. Recorded mass spectrum of protonated AlkiC4 (black bars) for $E/N = 70$ Td, corresponding to the highest sensibility for the protonated analyte signal detection (m/z 120) in absence of the RF device (a) and in the NO^+ ionization mode in presence of the RF ($E/N^* = 34$ Td) (b). The green thin line represents the expected molecular ion of the analyte. The intense signals depicted by the red thin bars represent the ionizing and matrix analytes at $m/z = 30$ (NO^+) and 46 (NO_2^+) and their isotopic abundance signals at $m/z = 31$ and $47, 48$ respectively. The signals corresponding to the water clusters ($\text{H}_3\text{O}^+(\text{H}_2\text{O})_n$) at m/z 37 and 55 are systematically erased for simplification.

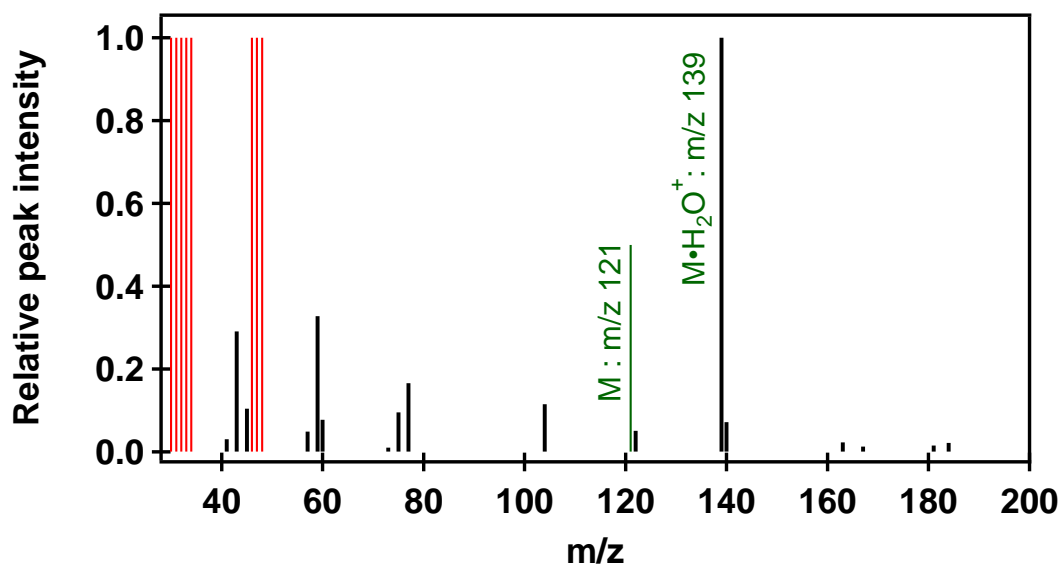


Figure S3. The recorded mass spectrum of 1OH3C3 (black bars) at the lowest extent of fragmentation ($E/N^* = 45$ Td) in the H_3O^+ ionization mode under the influence of the RF mode.

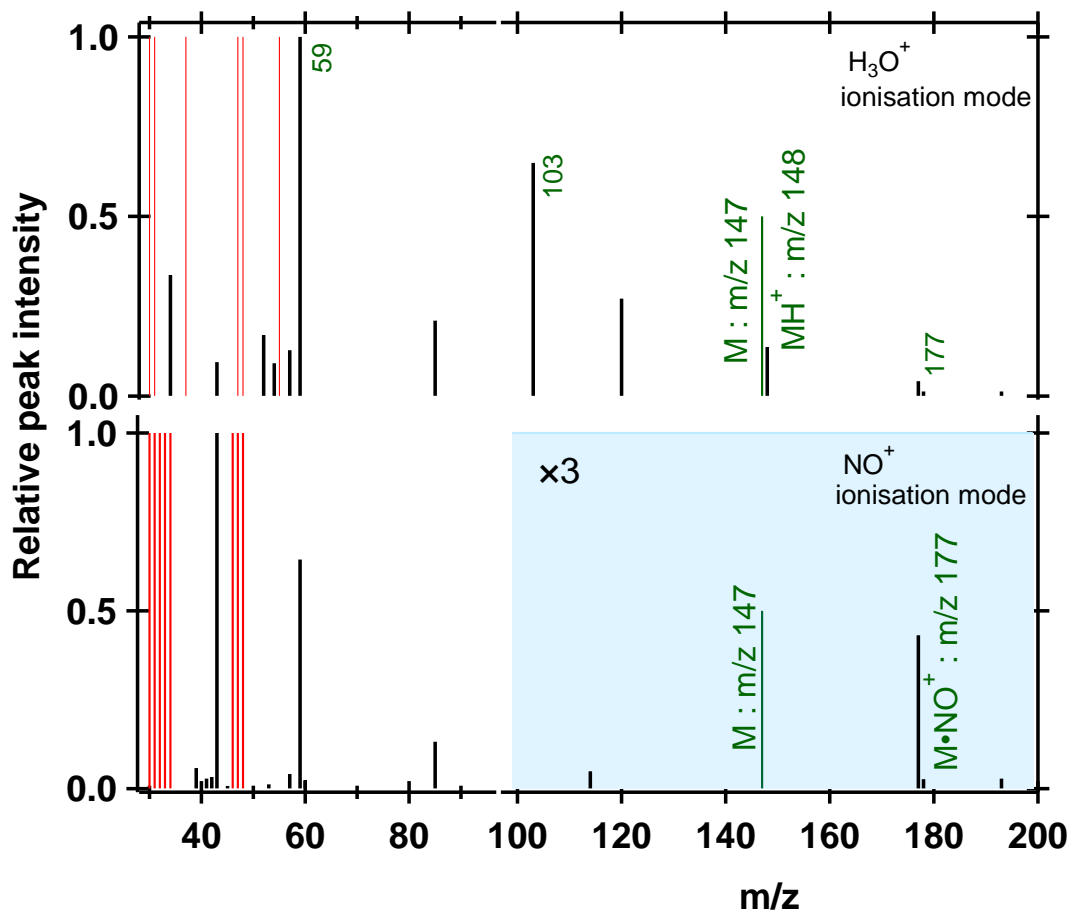


Figure S4. Recorded mass spectrum of protonated KnC5 (black bars) for $E/N = 75$ Td, corresponding to the highest sensibility for the protonated analyte signal detection (m/z 148) in absence of the RF device (a) and in the NO^+ ionization mode in presence of the RF ($E/N^* = 36$ Td) (b). The green thin line represents the expected molecular ion of the analyte. The intense signals depicted by the red thin bars represent the ionizing and matrix analytes at $m/z = 30$ (NO^+) and 46 (NO_2^+) and their isotopic abundance signals at $m/z = 31$ and 47, 48 respectively. The signals corresponding to the water clusters ($\text{H}_3\text{O}^+(\text{H}_2\text{O})_n$ at m/z 37 and 55 are systematically erased for simplification.

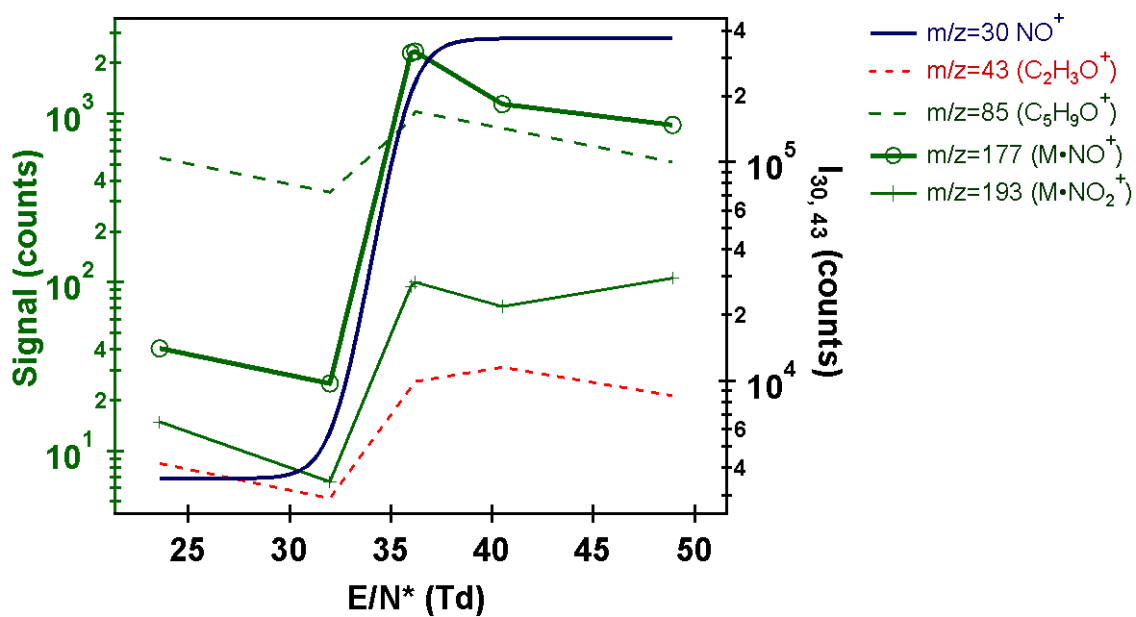


Figure S5. The RF mode behavior of typical NO^+ ionization signals of keto-nitrates (KnC5) under the E/N ratio influence (left axis). Typical distribution of the NO^+ ions and $\text{C}_2\text{H}_3\text{O}^+$ fragment into the given E/N interval (right axis).

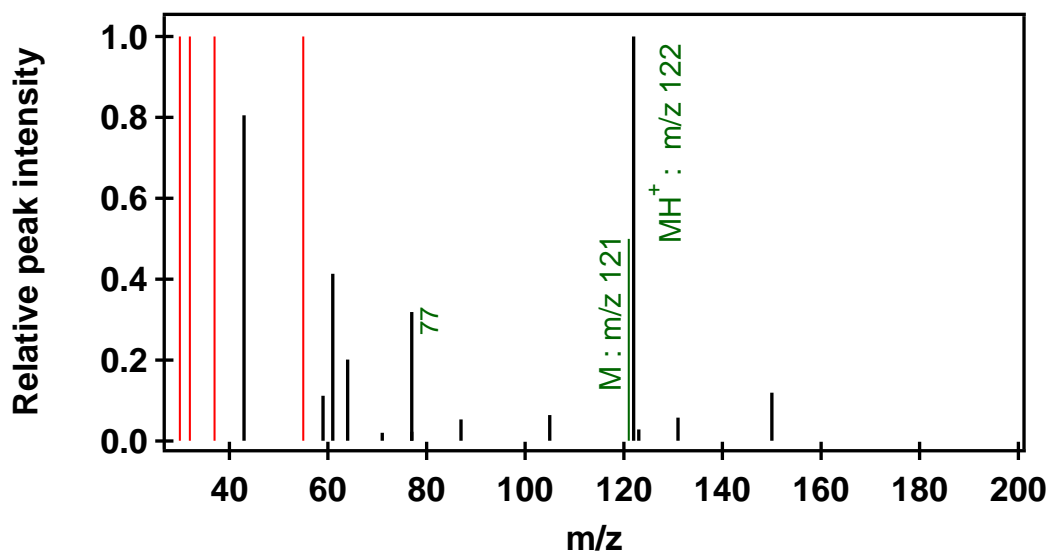


Figure S6. Recorded mass spectrum of protonated PAN (black bars) for $E/N = 85$ Td, corresponding to the highest sensibility of the m/z 122 signal detection.